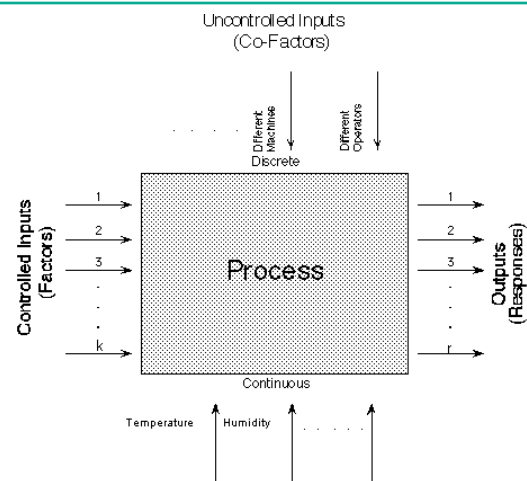
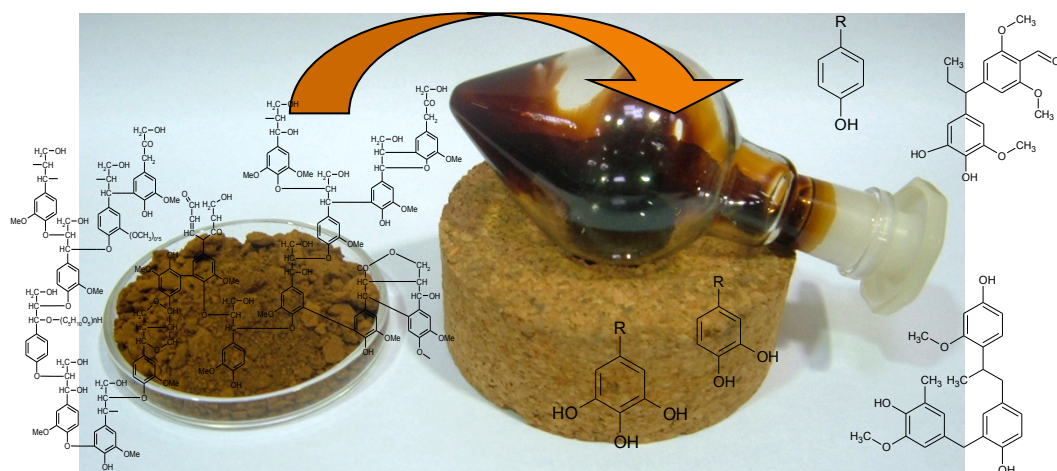


# Lignin to aromatic compounds. The Base Catalysed Degradation in continuous reactors - a tentative review

**Detlef Schmiedl<sup>1</sup>**, Gerd Unkelbach<sup>2</sup>, Sarah Böringer<sup>1</sup>, Detlef Rückert<sup>1</sup>, Rainer Schweppe<sup>1</sup>

<sup>1</sup>) Fraunhofer Institute for Chemical Technology ICT (Fh-ICT, Pfinztal, Germany),

<sup>2</sup>) Fraunhofer Center for Chemical-Biotechnological Processes CBP (Fh-CBP, Leuna, Germany)



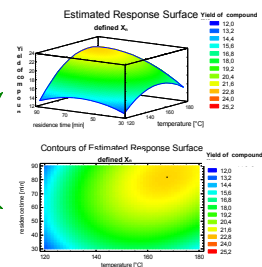
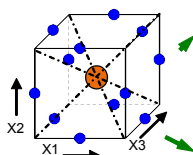
Factors  
X1 (e.g. Temperature)  
X2 (e.g.  $C_{NaOH}$ )  
X3 (e.g. residence time)

Levels

-1; 0; +1

-1; 0; +1

-1; 0; +1



# Biomass - A new feedstock for the 21. century ?

coal

(> 1950)

crude oil / natural gas

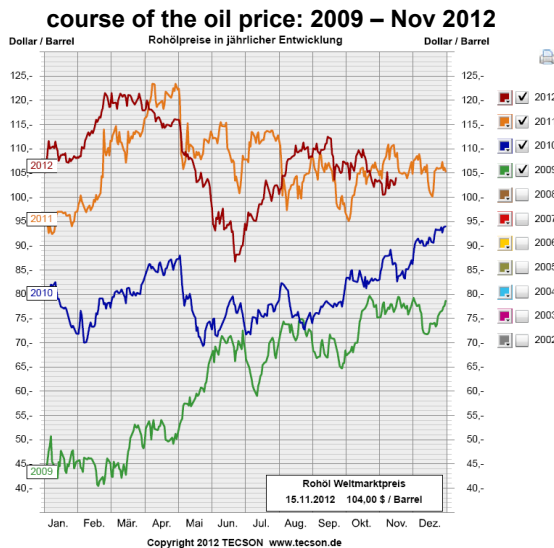
biomass ?



From coal to crude oil - new efficient techniques have been developed for an economic production of chemicals



For the shift from oil & gas to bio mass based chemicals it has to be the same



	Annual Production	Price of Product	Price of functionalized Carbon (2010)
ethylene	110 Mio. t/a	890 €/t	1035 €/t C
propylene	75 Mio. t/a	850 €/t	988 €/t C
benzene	45 Mio. t/a	830 €/t	902 €/t C
cellulose	320 Mio. t/a	500 €/t	1250 €/t C
starch	55 Mio. t/a	250 €/t	625 €/t C
sugar	143 Mio. t/a	300 €/t	750 €/t C
ethanol	36 Mio. t/a	365 €/t	700 €/t C
<b>lignin</b>	<b>?</b>	<b>?</b>	<b>?</b>

# Utilizing lignocellulose biomass

Fractionation of hardwood by a modified  
Organosolv-process

Pilot project “German lignocellulose bio refinery”

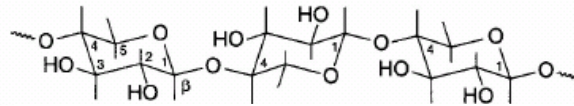
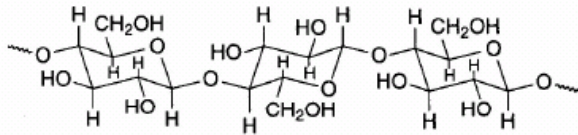


## Cellulose

40-55%

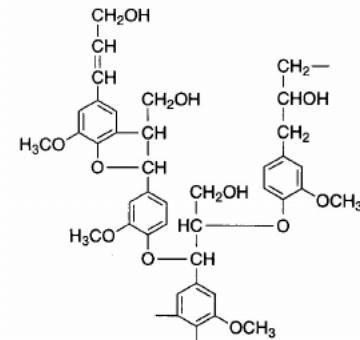
## Hemicellulose

15-35%



## Lignin

28-31% (soft wood);  
18-25% (hard wood)

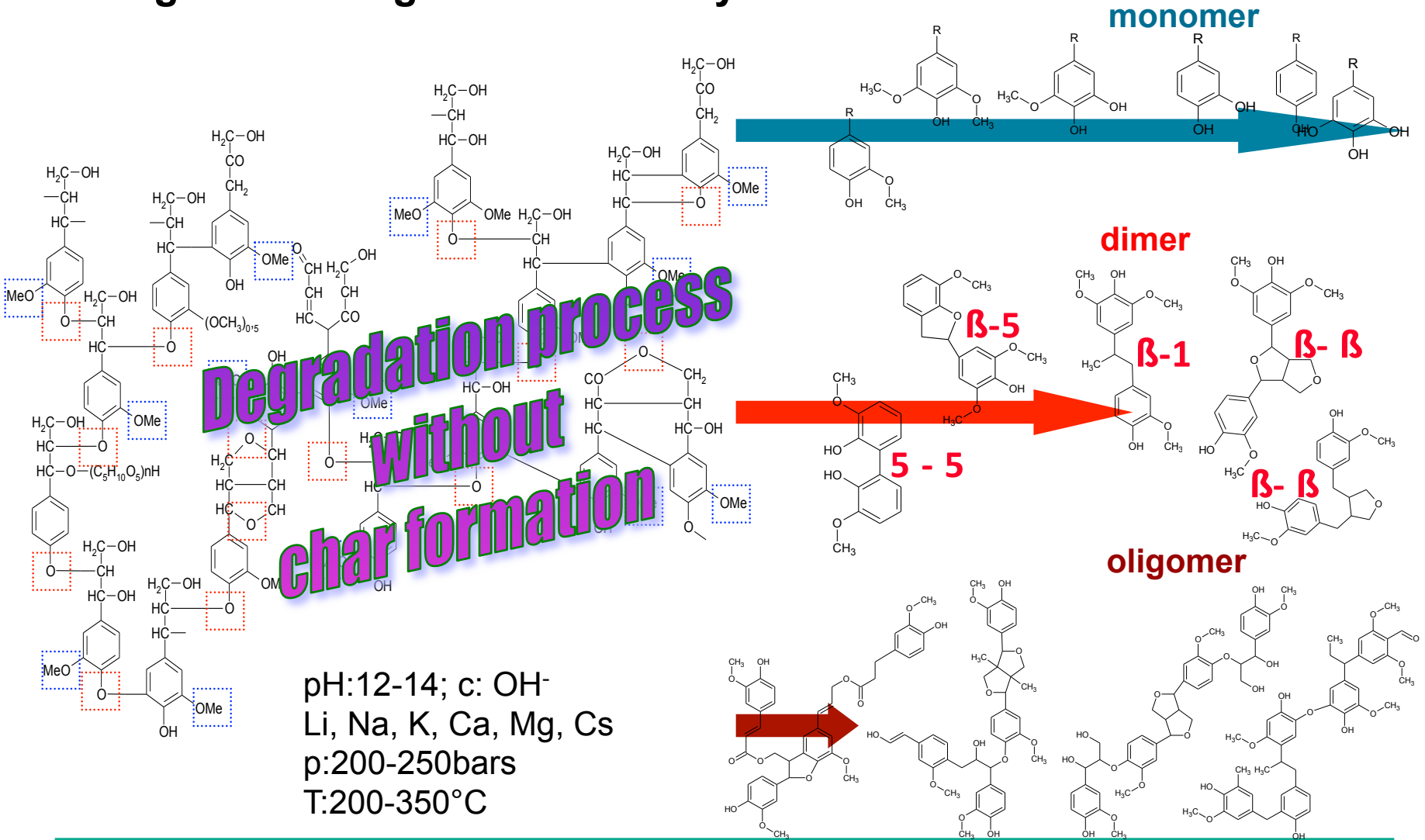


Hydrolysis results in  $C_6$ - and  $C_5$ -sugars as  
feedstock for fermentation or chemical  
reactions

Not yet an “easy” continuously  
degradation process for Lignin

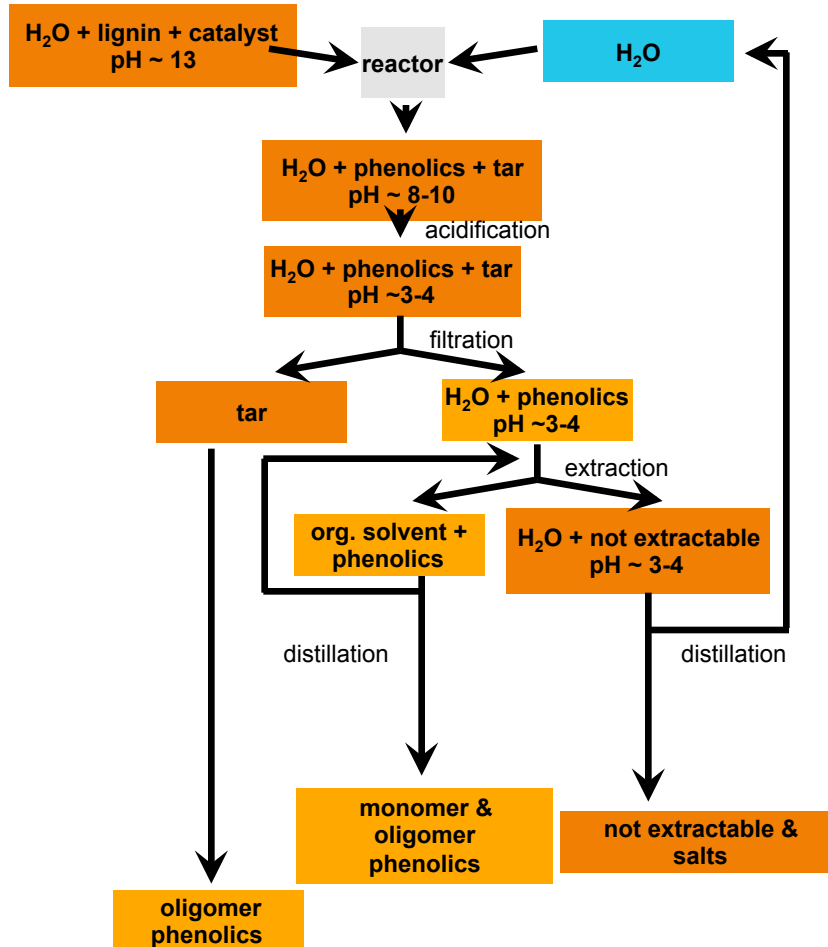
Only material applications  
(formaldehyde resins, PU foams, casting  
resins)

# Chemical principle of an alkaline hydrolysis process (BCD) on Lignin for the generation of oxy aromatics

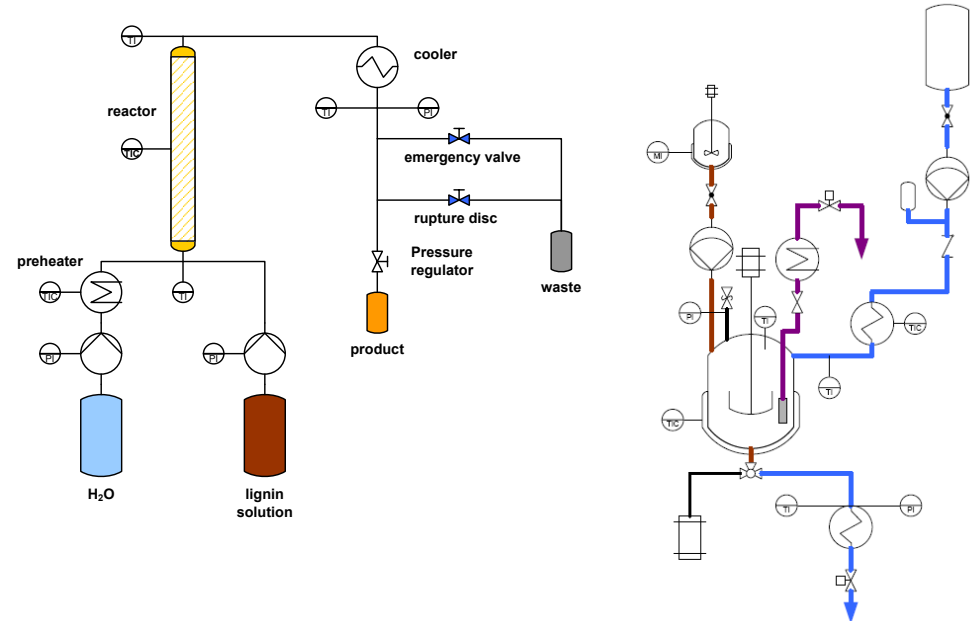


# Engineering principles of the BCD-process (Reaction & Separation Techniques)

BCD procedure & down stream



General flow diagram of a plug flow reactor & a continuously stirred tank reactor



T: 200-350°C,  
 p: up to 250 bar,  
 τ: 35sec to 15min dwell time  
 10 % Lignin

# Structural properties of lignin (botanical origin/incidence of bond type based on 100C<sub>9</sub> units)

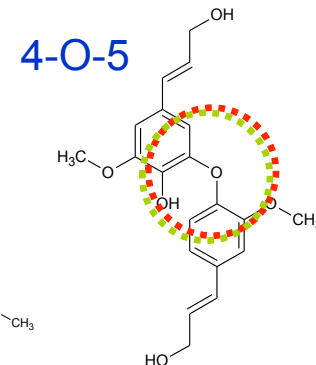
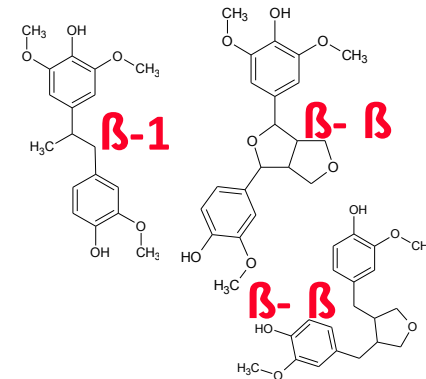
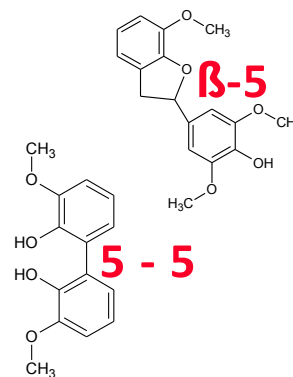
bond type	dimeric structural type	softwood (spruce) C <sub>9</sub> units of 100	hardwood (beech) C <sub>9</sub> units of 100
$\beta$ -O-4	$\beta$ -arylether	48	65
$\alpha$ -O-4	benzyl ether	6 – 8	
$\beta$ -5	phenyl ether		6
5-5	biphenyl	10 – 11	2
4-O-5	4-phenylether	4	2
$\beta$ -1	1,2-phenylene	7	15
$\beta$ - $\beta$	THF or resinol type	2	2 + 5

1/3 are alkyl bridges connecting aromatic rings - limitations for hydrolysis

Ref.: ERICKSON, M.; LARSSON, S.; MISCHKE, G.E. (1973)  
NIMZ, H.; LÜDEMANN, H.D. (1974)

## Structural properties of lignin

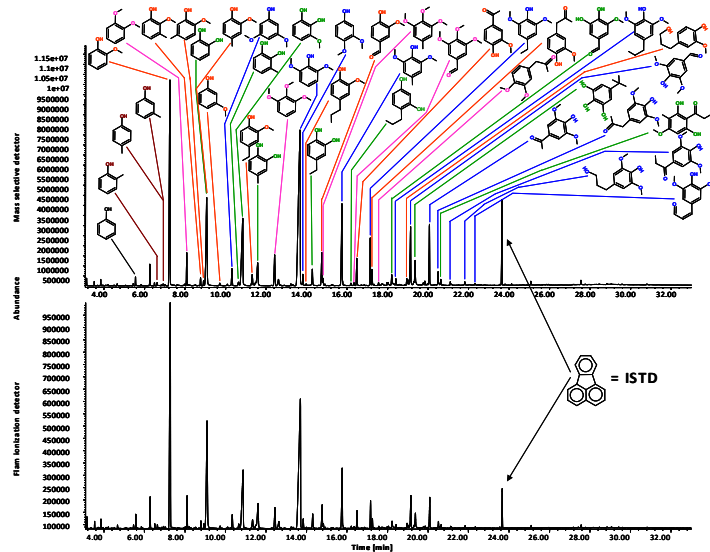
dimer & oligomer composition  
& functionality of BCD oil and tar





# Analytical methods

- Reactor water, extracts and filter-cakes were characterized by gravimetric analysis
- Elemental analysis (CHONS) on the oil and tar fractions
- Monomers were identified and quantified by GC-MSD/FID
- Reactor water was analyzed by HPLC RID (formic acid, acetic acid, methanol)
- Oligomeric oxyaromatics were analyzed by LC MSD Ion Trap XCT+ / SEC



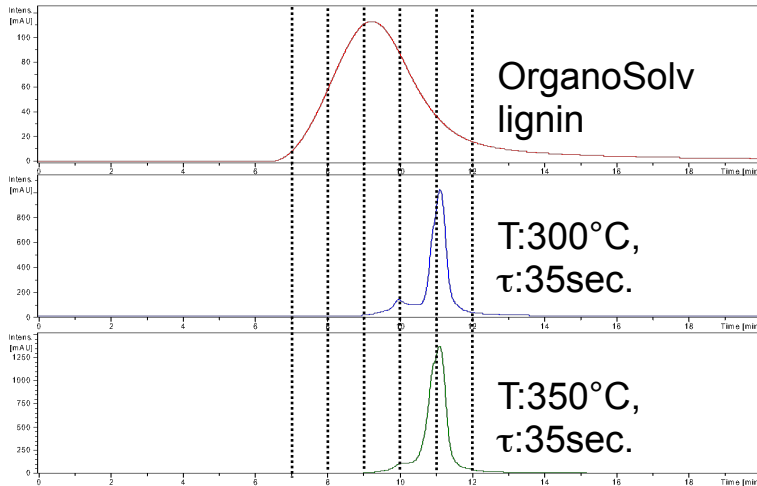
GC-MSD/FID analysis of monomer compounds in a BCD oil



# Analytical methods

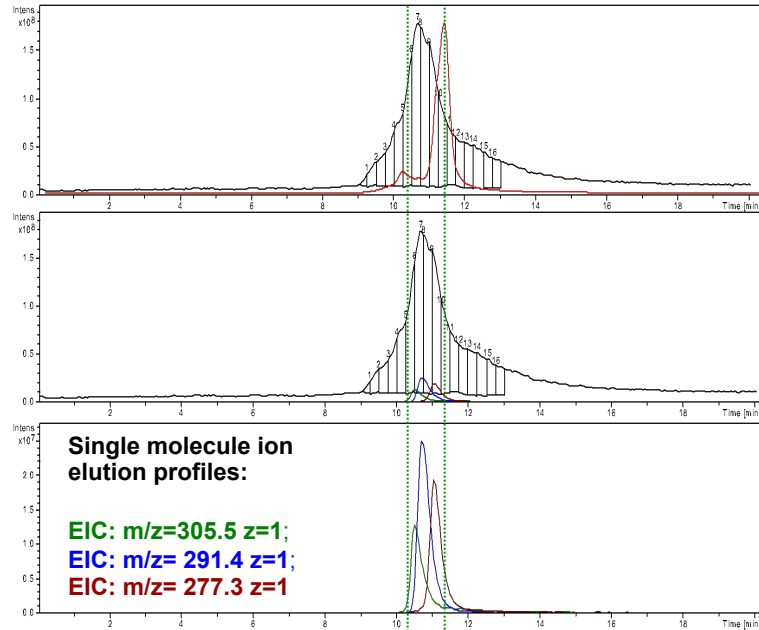
- Oligomeric oxyaromatics were analyzed on a SEC DAD MSD Ion Trap XCT+

## Elution profile - SEC / DAD (oil):



SEC of **lignin** (red), **oil L38**(blue), **oil L39**(green) on SDV micro, 4.6x30mm, 4.6x250mm, 3 $\mu$ , 50 Å (PSS Mainz, Germany), mobile phase: THF non-stabilized, flow rate: 0.250ml/min; DAD  $\lambda$ : 280nm  $\pm$  2nm, cell size: 5 $\mu$ l; Temp: 50°C, HPLC system: Agilent RR HT 1200SL.

## SEC / **DAD** / ESI neg. MSD Ion Trap (oil):



SEC of an oil (MSD Ion Trap, ESI negative:

**black:** TIC: scan:100-2200 amu;

**Dominant signals:**

**green:** EIC: m/z=305.5 z=1;

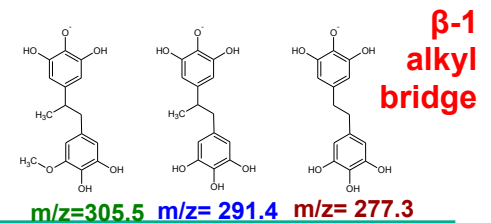
**blue:** EIC: m/z= 291.4 z=1;

**brown:** EIC: m/z= 277.3 z=1)

TIC: Total Ion Current

EIC: Extracted Ion Current

### Possible molecules

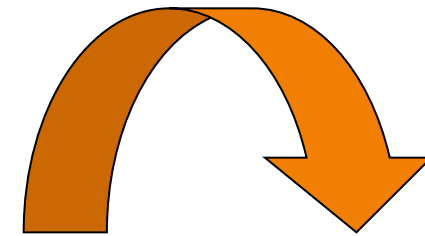
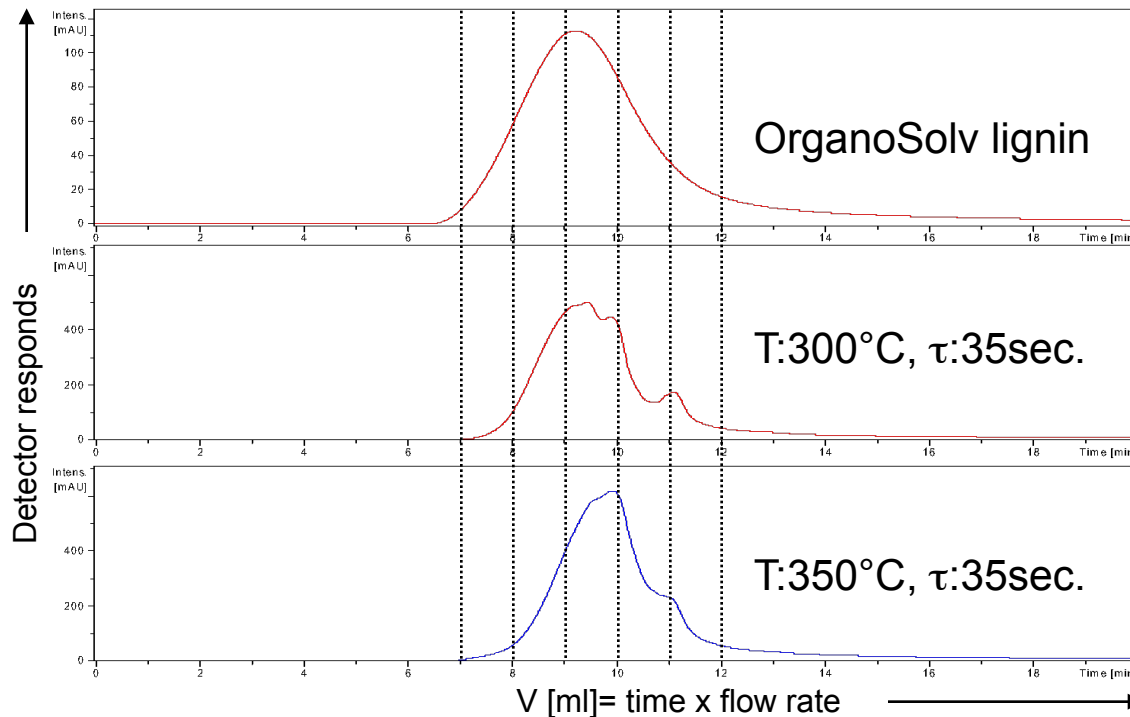




# Analytical methods

- Oligomeric oxyaromatics were analyzed on a SEC-DAD-MSD Ion Trap XCT+

## SEC / DAD (tar):



Process  
parameters

**Molecular  
Characteristics  
(Mw, Mn)  
of Tar**

SEC of **lignin** (upper), **tar L38** (red), **tar L39** (blue) on SDV micro, 4.6x30mm, 4.6x250mm, 3 $\mu$ , 50 Å (PSS Mainz, Germany), mobile phase: THF non-stabilized, flow rate: 0.250ml/min; DAD  $\lambda$ : 280nm  $\pm$  2nm, cell size: 5 $\mu$ l; Temp: 50°C, HPLC system: Agilent RR HT 1200SL.

# Structural properties of lignin (botanical origin/monomer type)

## Botanical source of lignin:

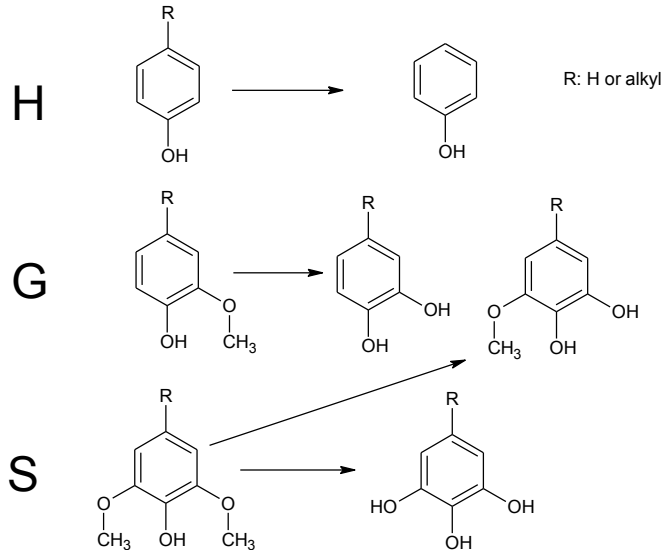
H/G/S-Lignin

G/S or S/G-Lignin (hw)

G-Lignin (sw)

annual plant

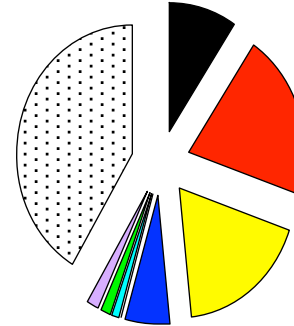
perennial plant



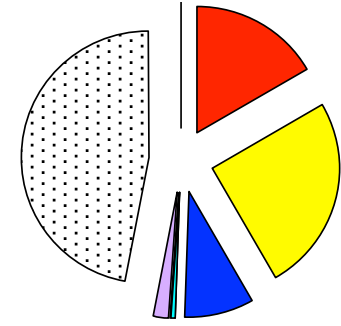
Structural properties

monomer composition of BCD oil

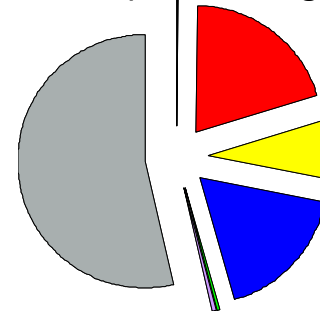
Crude oil (HGS lignin)



Crude oil (hard wood lignin)



Crude oil (technical lignin from a pulp mill)

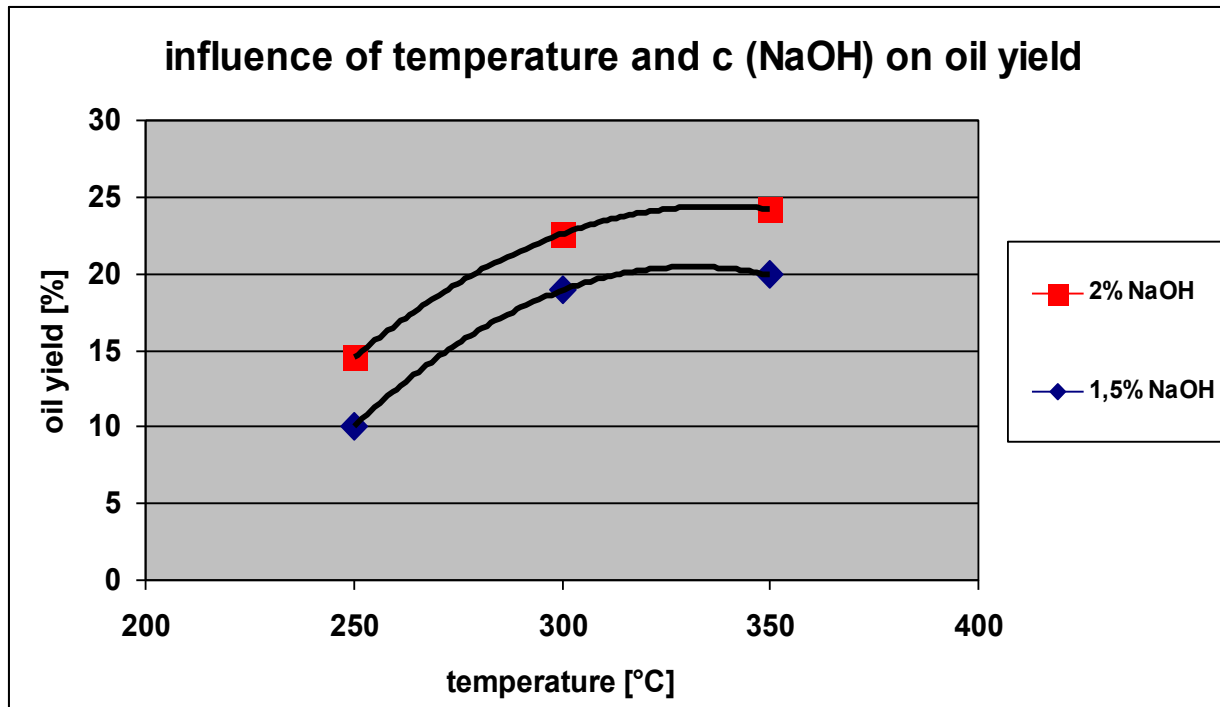


T:300°C,  
p:200-250bar,  
 $\tau$ =10min;  
ethyl acetate extract

- phenol
- guaiacol & monomethoxyphenols
- syringol & dimethoxyphenols
- o-hydroxyphenol derivatives
- m-hydroxyphenol derivatives
- benzene; 1,2,3- triol derivatives
- alkylphenols
- di- & trimethoxy-alkylbenzenes; alkylbenzenes
- unknown & oligomers

# Effect of Temperature & [NaOH] on oil yield

const.  $\tau=600\text{s}$ ,  $p=250\text{bar}$ ; ethyl acetate extracted; BCD on beech lignin

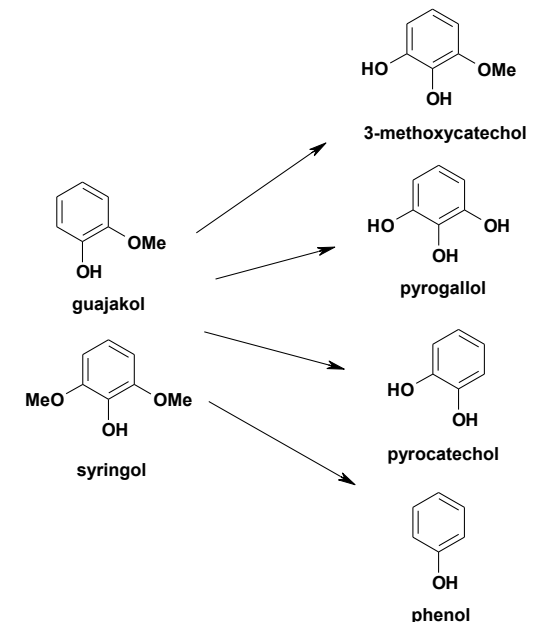


- Highest oil yields (ethyl acetate extracts) at 350°C with 2% NaOH.
- Adding more base has no significant effect on oil yield / more acid is needed during downstream processing of the reactor water

# Effect of Temperature on oil composition

const.  $\tau=600\text{s}$ ,  $p=250\text{bar}$ ; ethyl acetate extracted, BCD on beech lignin

Groups of compounds	wt.-% compared to oil		
	250°C	300°C	350°C
Phenol	0,1	0,1	0,4
Guaiacol & mono-methoxy phenol derivatives	8,8	16,4	22,9
Syringol & di-methoxy phenol derivatives	24,6	25,0	4,4
o-Hydroxy phenol derivatives	0,6	9,1	27,9
m-Hydroxyphenol derivatives	0,0	0,0	0,6
Benzene 1,2,3 –triol derivatives	0,0	0,5	0,8
Alkyl-phenol derivatives	0,0	0,0	0,4
Di- & Tri-methoxy-alkyl-benzenes, alkyl-benzenes	0,4	1,8	1,5
Unknown & Oligomer compounds	65,6	47,0	41,0

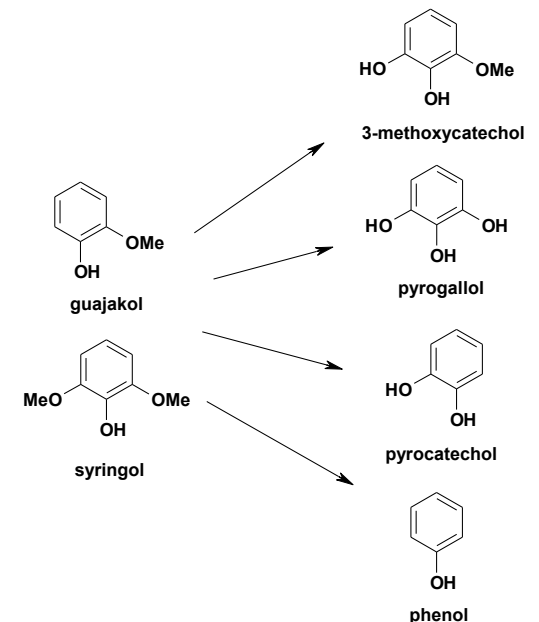


- Increasing reaction temperature increases hydroxy-functionalities in the oil ratification by experiments with model compounds
- The amount of oligomer compounds in the oil decreases by increasing T
- Monomers were identified and quantified by GC-MSD/FID on column HP-5ms

# Effect of residence time on oil composition

const.  $T=300^{\circ}\text{C}$ ,  $p=250\text{bar}$ ; ethyl acetate extracted; BCD on beech lignin

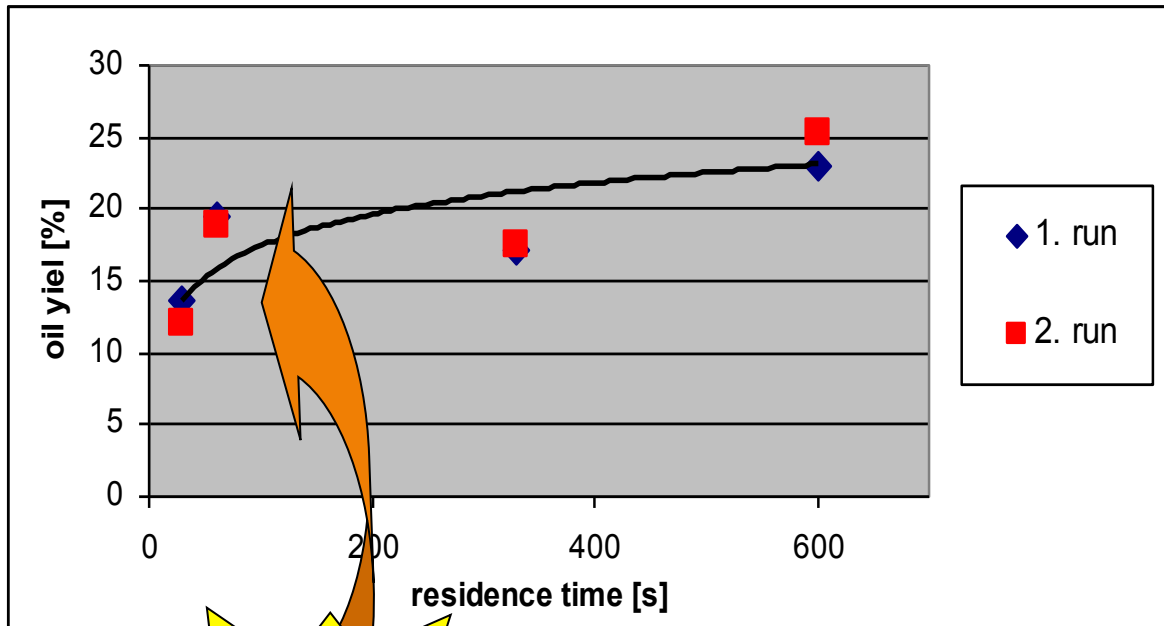
Groups of compounds	wt.-% compared to oil		
	300s	600s	900s
<b>Phenol</b>	<b>0,2</b>	<b>0,1</b>	<b>0,2</b>
<b>Guaiacol &amp; mono-methoxy phenol derivatives</b>	<b>17,4</b>	<b>16,4</b>	<b>14,4</b>
<b>Syringol &amp; di-methoxy phenol derivatives</b>	<b>32,7</b>	<b>25,0</b>	<b>23,6</b>
<b>o-Hydroxy phenol derivatives</b>	<b>6,2</b>	<b>9,1</b>	<b>15,8</b>
m-Hydroxyphenol derivatives	0,0	0,0	0,1
Benzene 1,2,3 –triol derivatives	0,6	0,5	0,9
Alkyl-phenol derivatives	0,1	0,0	0,2
Di- & Tri-methoxy-alkyl-benzenes, alkyl-benzenes	1,4	1,8	2,2
Unknown & Oligomer compounds	41,4	47,0	42,7



- Increasing residence time slightly increases hydroxy-functionalities in the oil ratification by experiments with model compounds
- The amount of oligomer compounds in the oil didn't change
- Monomers were identified and quantified by GC-MSD/FID on column HP-5ms

# Effect of residence time on oil yield

const.  $T=350^{\circ}\text{C}$ ,  $p=250\text{bar}$ ; ethyl acetate extracted; BCD on beech lignin



at 60s oil yields  
around 20%!

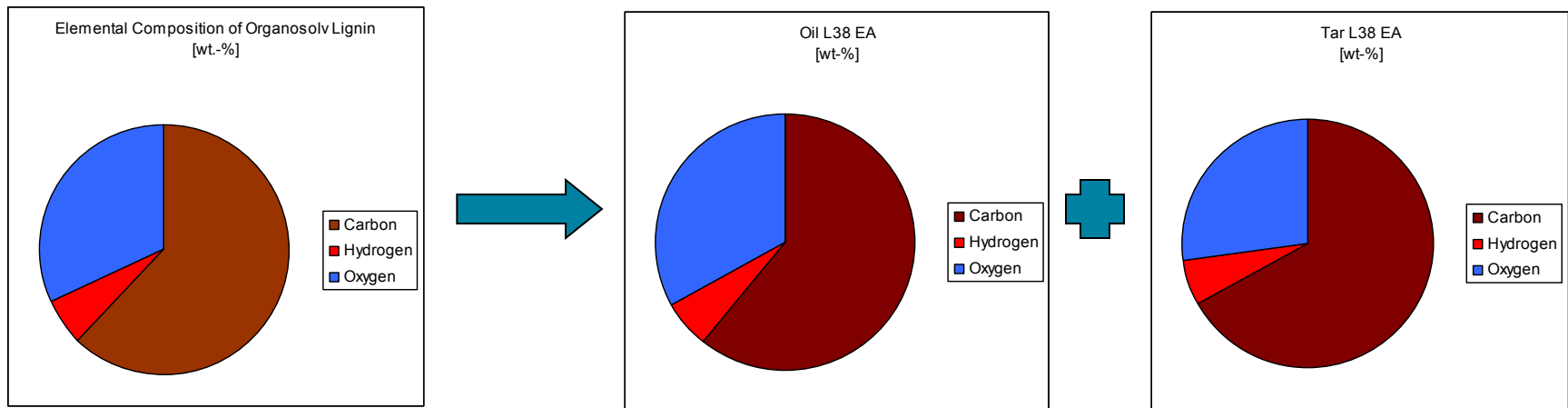
Investigations of the depolymerisation reactions in the first 2 min are very interesting

→ Studies with organosolv beech wood lignin and model compounds / detailed analytics is necessary !

# Carbon balance of BCD products

T=300 & 350°C, p=250bar,  $\tau$ =35s; ethyl acetate extracted; BCD on beech lignin

Fractions / phases	300°C	350°C
Oil	10.5 % Carbon of Lignin	18.0% Carbon of Lignin
Tar	73.2% Carbon of Lignin	56.8% Carbon of Lignin
formic acid, acetic acid, methanol in reactor water	2,5% Carbon of Lignin	4.4% Carbon of Lignin
Carbon lost via gaseous phase / downstream pro.	13.8% Carbon of Lignin	20.8% Carbon of Lignin



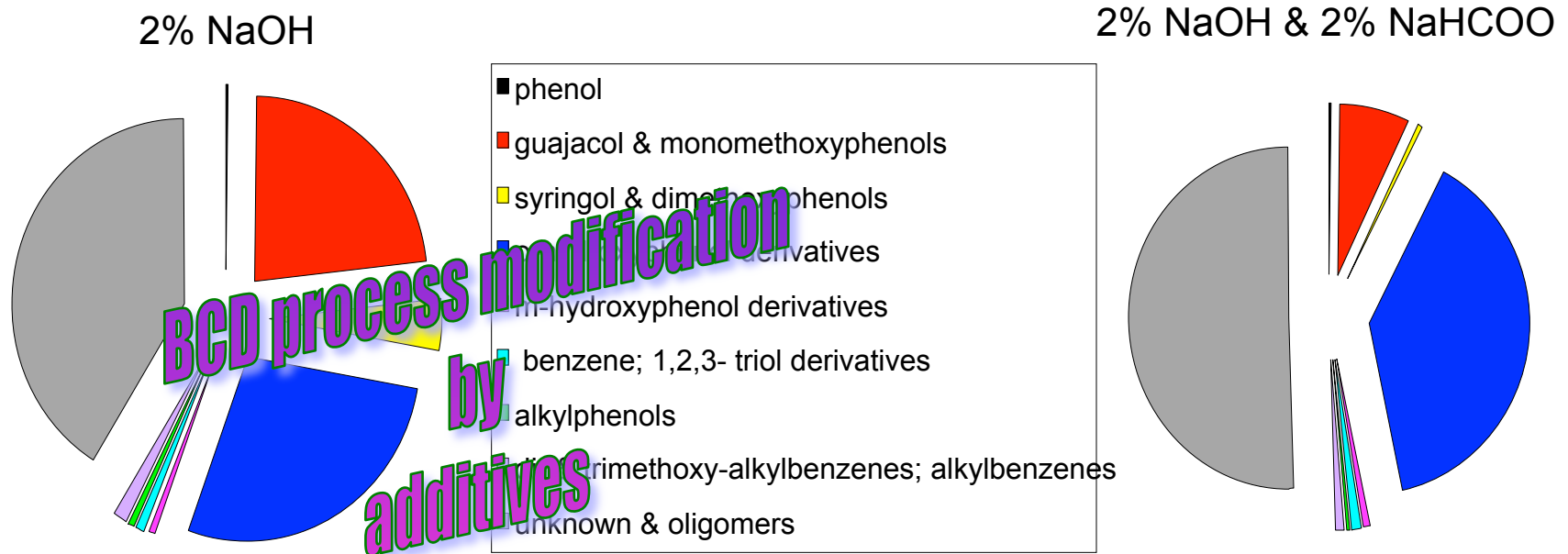
Carbon is the only element which is distributed during BCD !

Fractions are analyzed by HPLC-MS/RID/DAD and elemental analysis



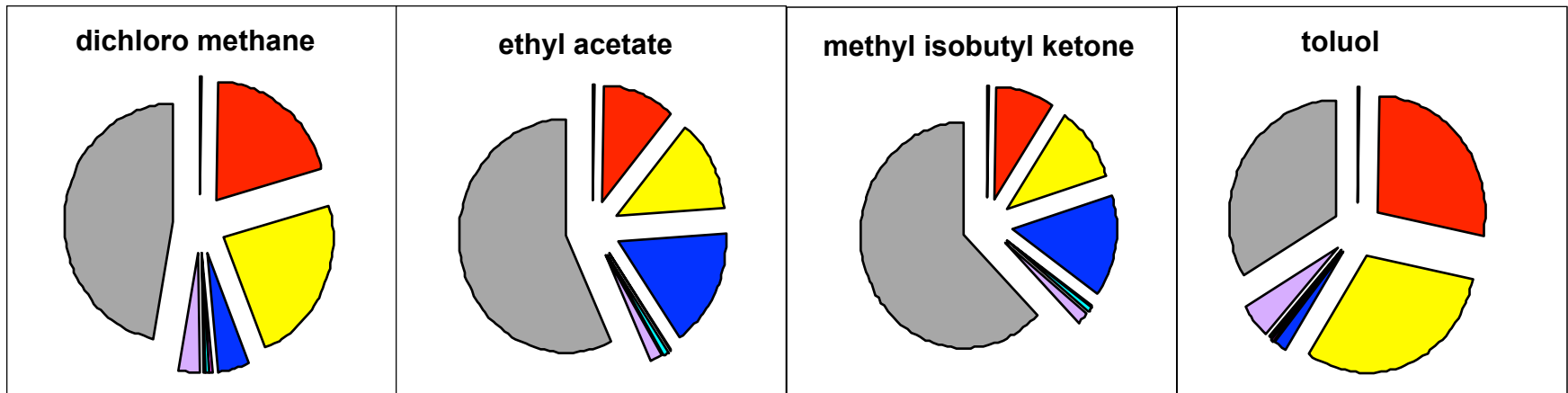
# BCD on beech lignin under reductive conditions / the Hydrogen donor effect

T=350°C, p=250bar,  $\tau$ =600sec; ethyl acetate extracted; BCD on beech lignin



Adding formate for *in situ* H<sub>2</sub>-generation decreases  
the oil yield from 25% to 19%,  
but produces more **catechols** derivatives.

# Compound distribution of oil depends also on extraction solvent



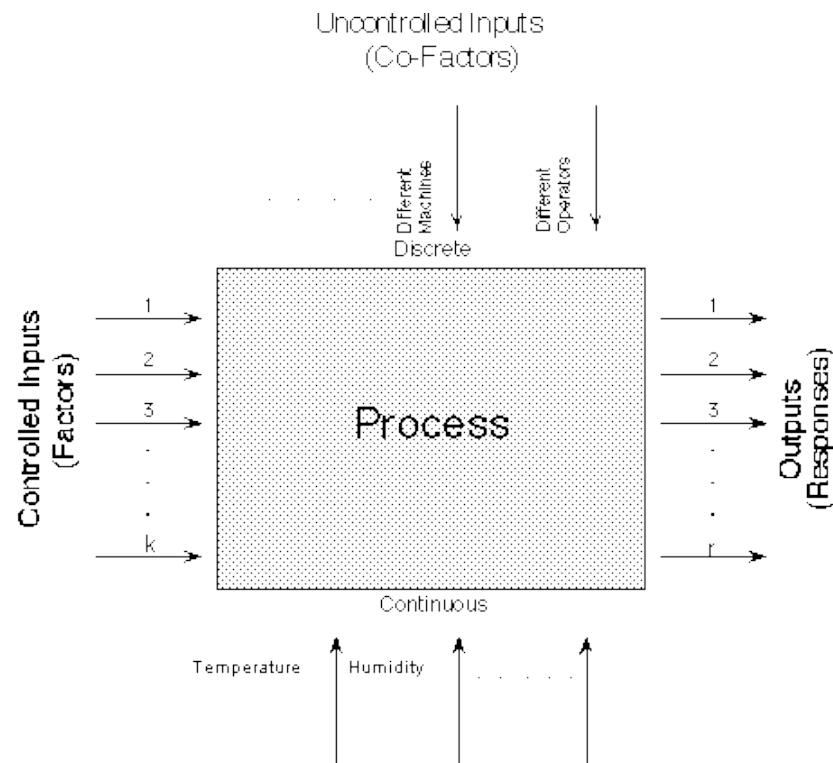
	wt.-% compared to oil			
	dichloro methane	ethyl acetate	methyl isobutyl ketone	toluol
<b>Phenol</b>	0,3	0,2	0,2	0,2
<b>Guaiacol &amp; mono-methoxy phenol deriv.</b>	20,2	10,9	9,1	27,8
<b>Syringol &amp; di-methoxy phenol derivatives</b>	23,4	12,7	10,9	31,1
<b>o-Hydroxy phenol derivatives</b>	4,6	16,5	14,7	2,1
m-Hydroxyphenol derivatives	0,4	0,3	0,2	0,3
Benzene 1,2,3 –triol derivatives	0,5	0,7	0,6	0,0
Alkyl-phenol derivatives	0,3	0,2	0,2	0,4
Di-&Trimethoxy-alkyl-benz., alkyl-benzenes	3,2	1,7	1,6	4,8
<b>Oligomere compounds</b>	47,1	56,9	62,5	33,4

<b>Oil mass [mg] extratced from 750g reactor water</b>	<b>550</b>	<b>1070</b>	<b>1310</b>	<b>420</b>
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## Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

The BCD process as “Black Box” process  
on Kraft-Lignin



A 'Black Box' Process Model Schematic

The Box-Behnken design

TABLE 3.24 Structural Comparisons of CCC (CCI), CCF, and Box-Behnken Designs for Three Factors

CCC (CCI)				CCF				Box-Behnken			
Rep	X1	X2	X3	Rep	X1	X2	X3	Rep	X1	X2	X3
1	-1	-1	-1	1	-1	-1	-1	1	-1	-1	0
1	+1	-1	-1	1	+1	-1	-1	1	+1	-1	0
1	-1	+1	-1	1	-1	+1	-1	1	-1	+1	0
1	+1	+1	-1	1	+1	+1	-1	1	+1	+1	0
1	-1	-1	+1	1	-1	-1	+1	1	-1	0	-1
1	+1	-1	+1	1	+1	-1	+1	1	+1	0	-1
1	-1	+1	+1	1	-1	+1	+1	1	-1	0	+1
1	+1	+1	+1	1	+1	+1	+1	1	+1	0	+1
1	-1.682	0	0	1	-1	0	0	1	0	-1	-1
1	1.682	0	0	1	+1	0	0	1	0	+1	-1
1	0	-1.682	0	1	0	-1	0	1	0	-1	+1
1	0	1.682	0	1	0	+1	0	1	0	+1	+1
1	0	0	-1.682	1	0	0	-1	3	0	0	0
1	0	0	1.682	1	0	0	+1				
6	0	0	0	6	0	0	0				
Total Runs = 20				Total Runs = 20				Total Runs = 15			

# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

## The Box-Behnken design

is an independent quadratic design (for the response surface methodology) based on combined statistical tests and quadratic equations.

In this design the treatment combinations (combination of Process parameters ( $X_n$ )) are at the **midpoints of edges (12)** of the process space and at the **center (3)**.

These designs are rotatable (or near rotatable) and require 3 levels of each factor:

### Factors

X1 (e.g. Temperature)

X2 (e.g.  $C_{NaOH}$ )

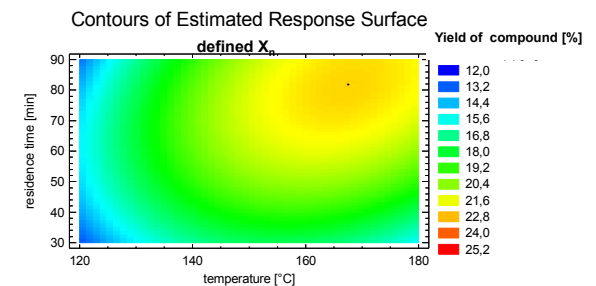
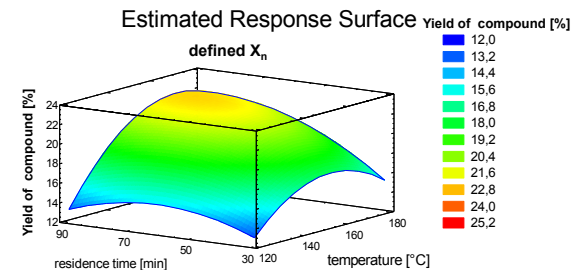
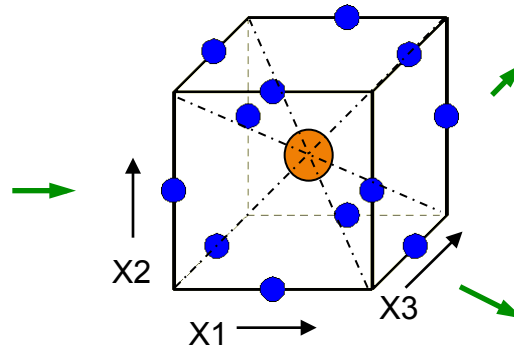
X3 (e.g. residence time)

### Levels

-1; 0; +1

-1; 0; +1

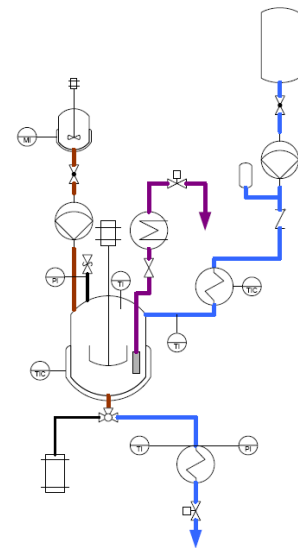
-1; 0; +1



## Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

### What are the goals of DoE in BCD of technical Lignin?

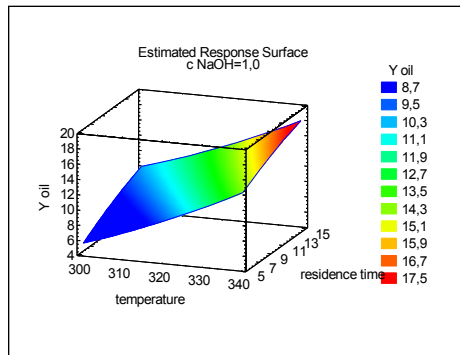
- Parametric studies (variable  $T$ ,  $\tau$ ,  $C_{\text{catalyst}}$ ) at const.  $p$ ,  $C_{\text{technical Lignin}}$   
(screening experiments to evaluate the reaction behavior)
- Effect of the variable process parameters/ controlled inputs ( $T$ ,  $\tau$ ,  $C_{\text{catalyst}}$ ) & Effect of a Co-factor (mineral content: 12wt.-%; ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ )) on:
  - Yield of Oil and Tar
  - Monomer composition of the Oil
  - Elemental composition of Lignin, Oil and Tar
  - Carbon balance / Carbon distribution over phases
  - Molecular weight of Tar



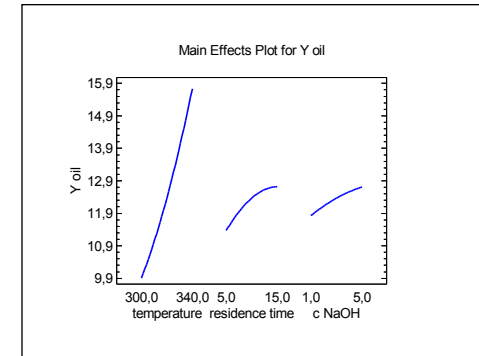
# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

[NaOH]

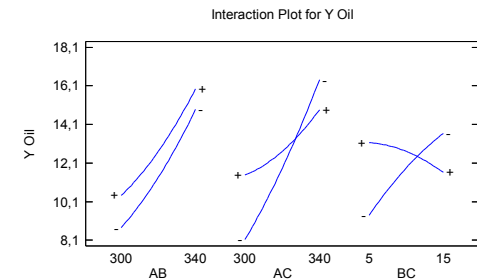
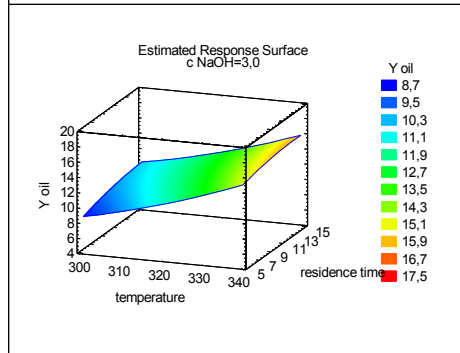
Low  
(1%)



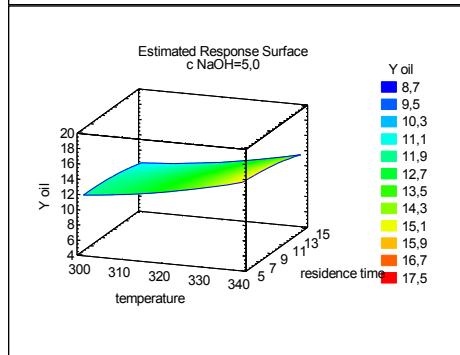
Effect of C<sub>NaOH</sub>  
on  
the oil Yield



Medium  
(3%)



High  
(5%)



Optimize Response

Goal: maximize Yield of Oil [wt.-%]

Estimated optimum value = 17,94 [wt.-%]

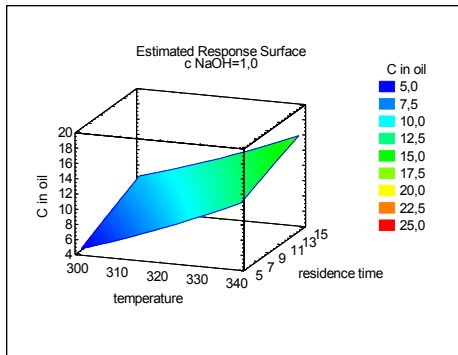
Correlation:  $R^2 = 95,44 \%$

Factor	Low	High	Optimum
Temperature [°C]	300,0	340,0	339,433
Dwell time [min]	5,0	15,0	15,0
C <sub>NaOH</sub> [%]	1,0	5,0	1,0

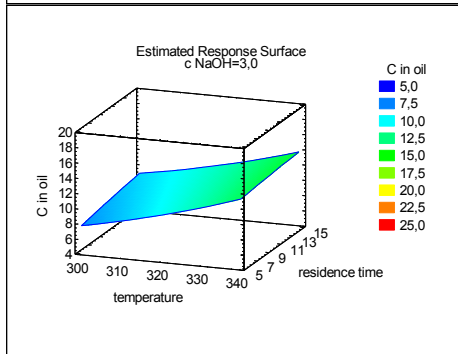
# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

[NaOH]

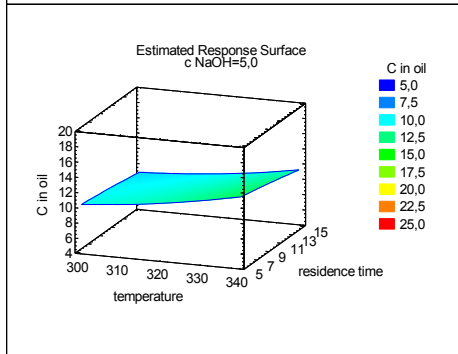
Low  
(1%)



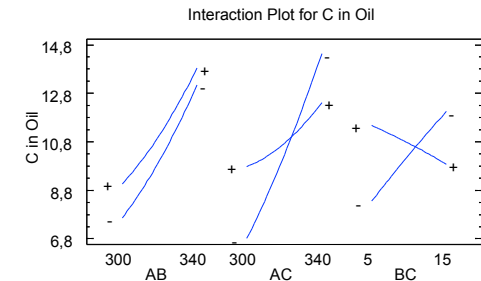
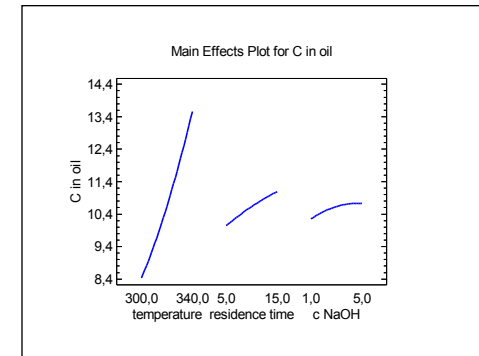
Medium  
(3%)



High  
(5%)



Effect of  $C_{NaOH}$   
on the  
relative  
Carbon content  
in oil



Optimize Response

Goal: maximize Carbon in Oil originating from Lignin [wt.-%] –

Conversion degree into Oil

Estimated optimum value = 15,91 [wt.-%]

Correlation:  $R^2 = 94,32 \%$

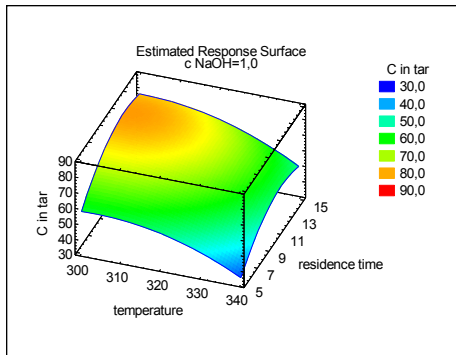
Factor	Low	High	Optimum
Temperature [°C]	300,0	340,0	339,437
Dwell time [min]	5,0	15,0	15,0
$C_{NaOH}$ [%]	1,0	5,0	1,0



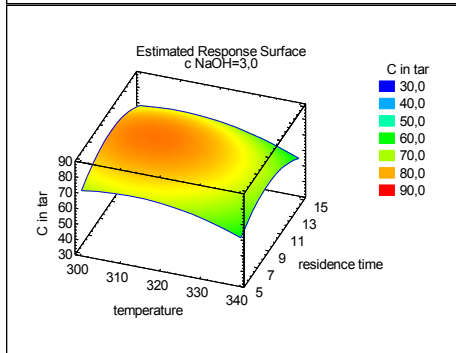
# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

[NaOH]

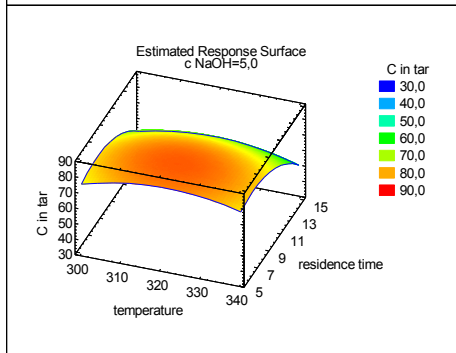
Low  
(1%)



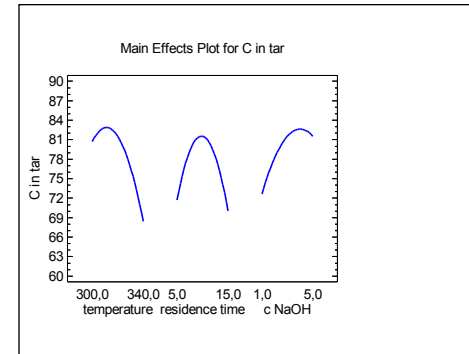
Medium  
(3%)



High  
(5%)



Effect of C<sub>NaOH</sub>  
on the  
relative  
Carbon content  
in Tar



Optimize Response

Goal: maximize Carbon in Tar originating from Lignin [wt.-%] –

Conversion degree into Tar

# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

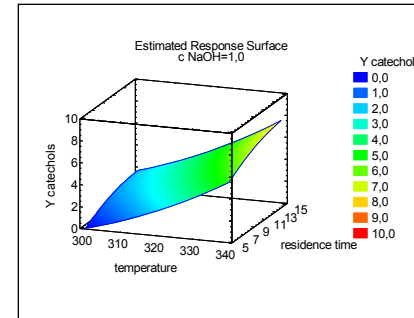
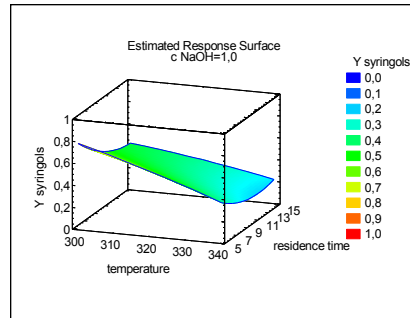
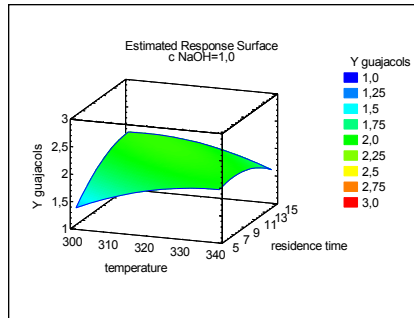
[NaOH]

Guaiacol type  
monomers

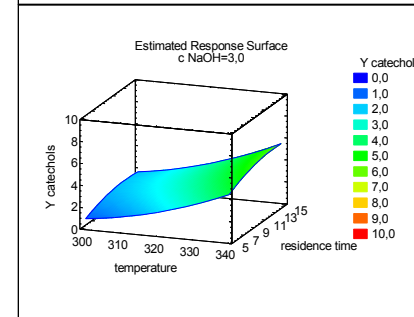
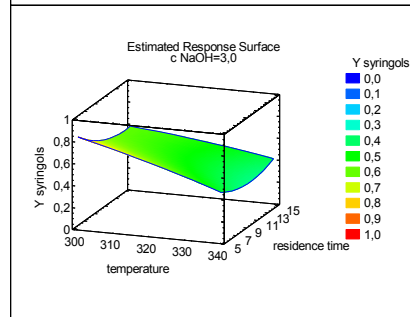
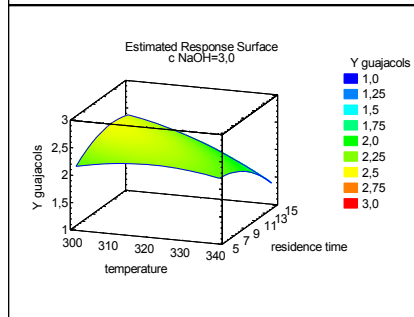
Syringol type  
monomers

Catechol type  
monomers

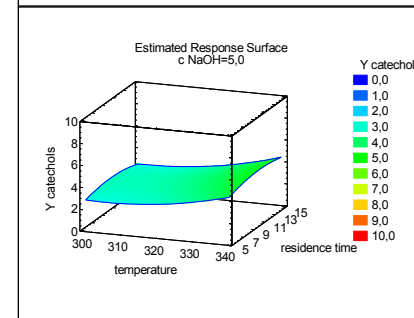
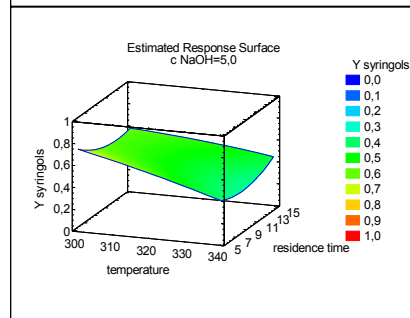
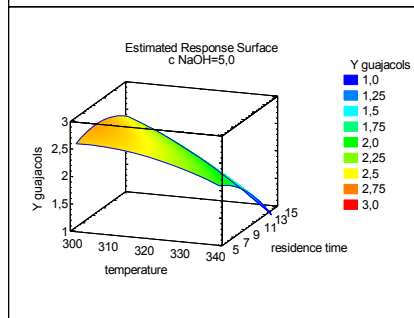
Low  
(1%)



Medium  
(3%)



High  
(5%)



**Effect of  
C<sub>NaOH</sub> on  
the Yield  
of selected  
monomer  
compounds  
in  
BCD oil**

# Part II. Lignin cleavage for the production of oxy aromatics/ technical Lignin from a pulp mill

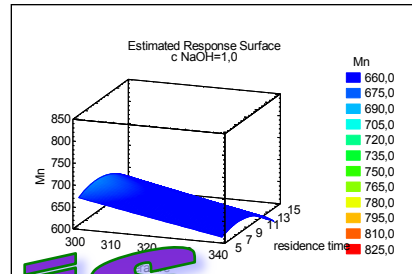
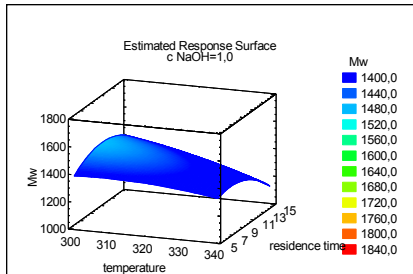
[NaOH]

Mw

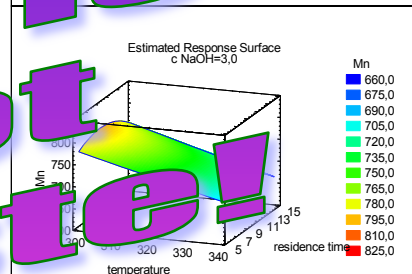
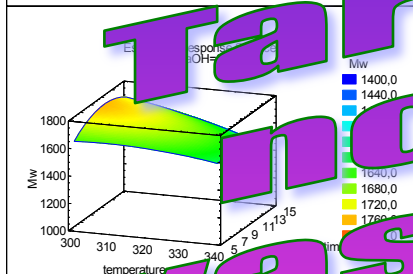
Mn

technical Lignin:  
Mw: 3592g/moles  
Mn: 1022g/moles  
PD: 3.54

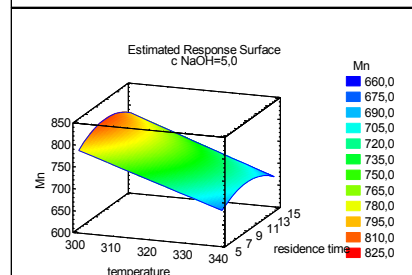
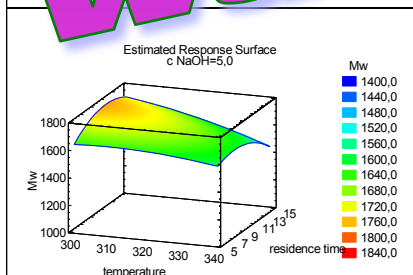
Low  
(1%)



Medium  
(3%)



High  
(5%)

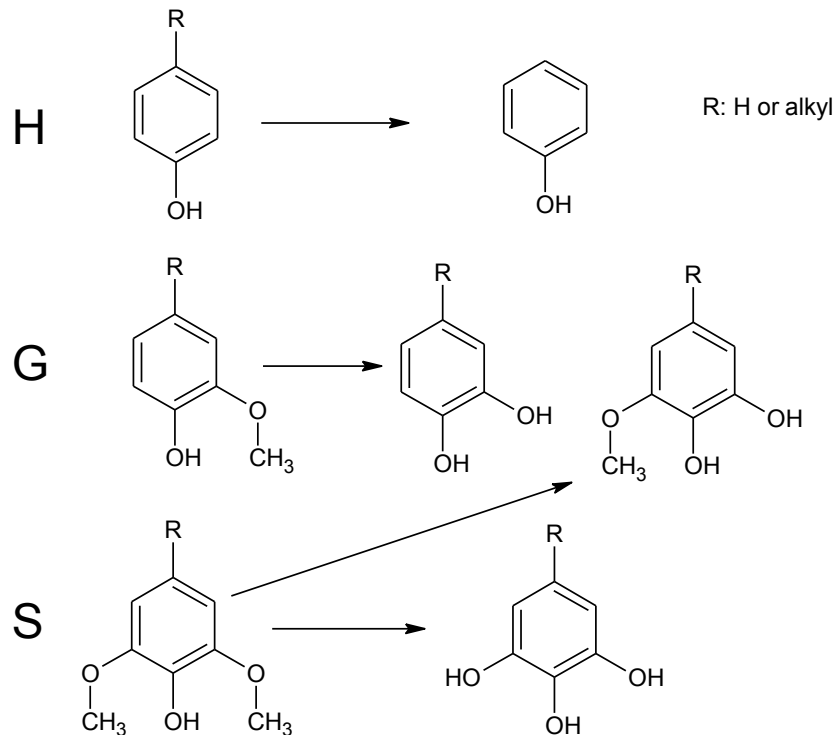


**Tar is not waste!**

Effect of  
 $C_{NaOH}$  on the  
molecular  
characteristic  
(Mw, Mn) of Tar

# Summary: Process factors / substrate factors effect during BCD on Lignin

- Generally process parameters & the type of lignin influence the yield, composition & properties of the fractions (tar , oil)



## process parameters:

T,  $\tau$ , p, catalyst

presence of additives

## Botanical source/ Pulping conditions:

HGS-Lignin (annual plants)

GS/SG-Lignin (hard wood)

G-Lignin (soft wood)

Mixtures

# Summary

- Oil yields are around 20 wt% in short residence times by BCD on OrganoSolv Lignin (beech)
- The yield of water insoluble (pH=3) oligomeric compounds (tar) is approx. 50wt% of Lignin
- Addition of sodium formiate favors the cleavage of aryl-methyl-ether bonds → increase of hydroxy functionalities in oil and tar
- Depending on the process parameters (T, p, residence time, catalyst) and the downstream processing a defined product distribution can be reached
- The downstream processing / extraction has to be optimized
- Only combined processes (- subsequent bond splitting by hydro cracking, pyrolysis or oxidation of Tar) will be successful for high monomer aromatic yields
- BCD on Lignin is also a pretreatment step for biotechnological conversion routes.

## Acknowledgment:

Agency for Renewable Resources (Ministry for Agriculture, Forestry, Fishery & Consumer protection), Germany, for financial support

Dow Chemical for cooperation

Nordic pulp mill & paper industry for cooperation and financial support

vTI, Hamburg, Germany (D. Meier (PhD)) for the references & advice

International Lignin Institute, (A. Abaecherli) for the Lignin of an annual plant

Colleagues of Fraunhofer ICT (Pfinztal, Germany) for the team work

## Contact

### **Detlef Schmiedl, PhD**

Fraunhofer Institute for Chemical Technology ICT  
Environmental Engineering  
Joseph-von-Fraunhofer-Str.7  
D-76327 Pfinztal  
Phone ++49(0)721/4640-747  
Fax ++49(0)721/4640-111  
E-mail: [detlef.schmiedl@ict.fraunhofer.de](mailto:detlef.schmiedl@ict.fraunhofer.de)  
<http://www.ict.fraunhofer.de>

### **Gerd Unkelbach**

Fraunhofer Center for Chemical-Biotechnological Processes CBP  
Am Haupttor (Bau 4310)  
06243 Leuna  
Phone ++49 (0)721/4640-605  
E-mail: [gerd.unkelbach@cbp.fraunhofer.de](mailto:gerd.unkelbach@cbp.fraunhofer.de)  
<http://www.cbp.fraunhofer.de>

**Thank you for your attention!**